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### SYNTHESIS OF SYMMETRICAL AND UNSYMMETRICAL AMIDOPHOSPHITES

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## SYNTHESIS OF SYMMETRICAL AND UNSYMMETRICAL AMIDOPHOSPHITES

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A method is proposed for one-pot synthesis of either symmetrical or unsymmetrical diester amidophosphites, developed on the basis of tris-(N,N-dimethyl)-amide/or tris-(N,N-diethyl)-amide/of phosphorous acid, activated by iodine, as a new phosphorylating reagent.

Nine thiophosphate, phosphite and phosphite monoester and diester derivatives of 1-O-stearoyl-ethane-2-ol, cholesterol,  $\beta$ -sitosterol and R,S- $\alpha$ -tocopherol have been synthesized under mild conditions (20–75°C) and in high final yields (83–98%).

At present, various methods and reagents are useful in the phosphorylation of compounds containing a free hydroxyl group.<sup>1–3</sup> In this respect, the acyclic triamides of phosphorous acid are of special interest.<sup>4,5</sup> Being polyfunctional, however, these reagents cannot be subjected to stoichiometrical alcoholysis (or phenolysis) if monoester derivatives are desired. Consequently, three equivalents of the triamide are usually employed.<sup>5–7</sup> This excludes the possibility of obtaining unsymmetric diesters directly. Combined with the rather severe conditions needed (in some cases thermal treatment up to 140–160°C<sup>4,6</sup>), the scope of the phosphorylating reagents mentioned is considerably limited.

We now propose a method for one-pot synthesis of either symmetrical or unsymmetrical diester amidophosphites under mild conditions, by means of a new phosphorylating reagent developed on the basis of tris-(N,N-dialkyl)-amides of phosphorous acid activated by the addition of iodine at the optimum molar ratio 1.05:0.05.

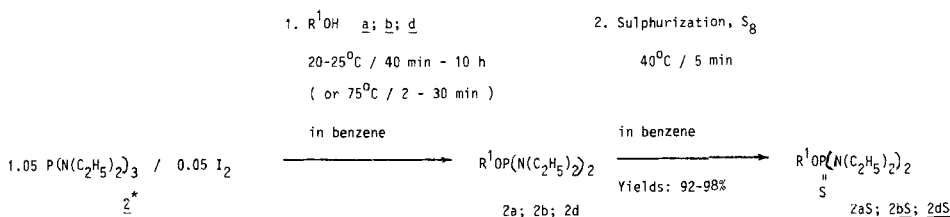
### RESULTS AND DISCUSSION

The tris-(N,N-dimethyl)-amide of phosphorous acid, **1** and the tris-(N,N-diethyl)-amide of phosphorous acid, **2**, both activated by iodine were used.

1-O-Stearoyl-ethane-2-ol, **a**, cholesterol, **b**,  $\beta$ -sitosterol, **c** and R,S- $\alpha$ -tocopherol, **d** were selected as model lipid substrates containing primary, secondary or sterically hindered phenolic hydroxyl functions.

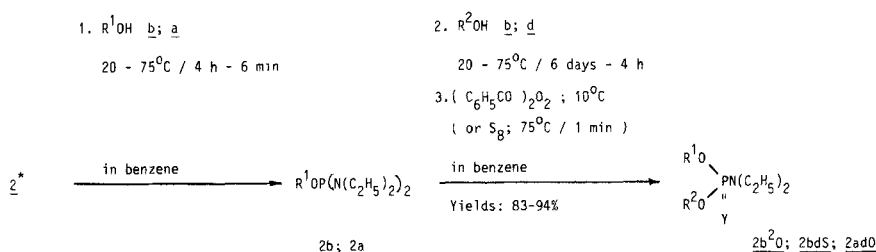
The reaction takes place either at room temperature or on moderate heating (70–75°C) in stoichiometric ratio with the substrates chosen, to give the respective monoester derivatives **2aS**, **2bS**, **2dS** in practically quantitative yield (Scheme 1):

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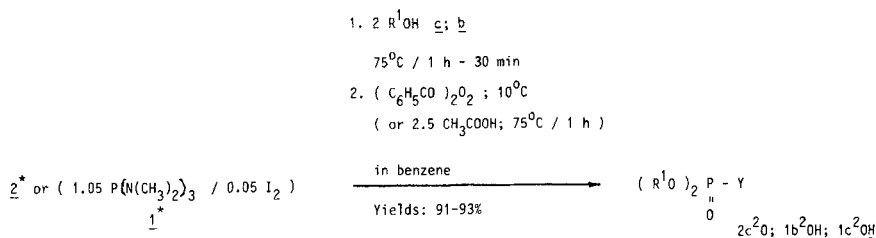
SCHEME 1 For **a**, **2a** and **2aS**:  $R^1 = 1\text{-O-Stearoyl}$ ethyl; **b**, **2b** and **2bS**:  $R^1 = \text{Cholesteryl}$ ; **d**, **2d** and **2dS**:  $R^1 = R,S\text{-}\alpha\text{-Tocopheryl}$ .

The next step, regarding the synthesis of symmetrical, **2b<sup>2</sup>O** or unsymmetrical, **2bdS**, **2adO** diesters, may be performed also in high yield by directly reacting the previously prepared monoester intermediate, **2b**, **2a** with the equivalent quantity of another substrate (Scheme 2):



SCHEME 2 For **2b<sup>2</sup>O**:  $R^1 = \text{Cholesteryl}$ ;  $R^2 = \text{Cholesteryl}$ ;  $Y = O$ : **2b<sup>2</sup>S**:  $R^1 = \text{Cholesteryl}$ ;  $R^2 = R,S\text{-}\alpha\text{-Tocopheryl}$ ;  $Y = S$ ; **2adO**:  $R^1 = 1\text{-O-Stearoyl}$ ethyl;  $R^2 = R,S\text{-}\alpha\text{-Tocopheryl}$ ;  $Y = O$ .

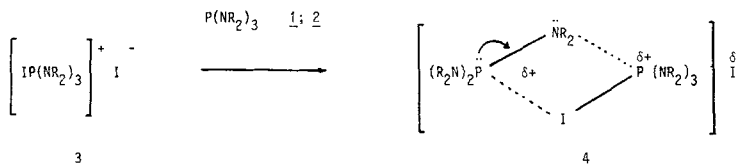
Symmetrical diester compounds may be obtained also, **2c<sup>2</sup>O**, by a one-step synthesis at molar ratio: reagent/substrate 1/2 (Scheme 3):



SCHEME 3 For **c** and **2c<sup>2</sup>O**:  $R^1 = \beta\text{-Sitosteryl}$ ;  $Y = N(C_2H_5)_2$ ; **1b<sup>2</sup>OH**:  $R^1 = \text{Cholesteryl}$ ;  $Y = H$ ; **1c<sup>2</sup>OH**:  $R^1 = \beta\text{-Sitosteryl}$ ;  $Y = H$ .

On the basis of the synthesized **P<sup>III</sup>** mono- or diester derivatives a wide range of additional transformations<sup>2,5,8</sup> could be performed. Thus, in all cases, the structures of the amidophosphites (Schemes 1-3) prepared were proven after convenient chemical transformations to thiophosphates, **2aS**, **2bS**, **2dS**, **2bdS**, phosphates, **2b<sup>2</sup>O**, **2adO**, **2c<sup>2</sup>O**, and phosphites, **1b<sup>2</sup>OH**, **1c<sup>2</sup>OH**.

It could be suggested that the demonstrated high selectivity of the reagent is due to the action of an initial quasi-phosphonium compound,<sup>9</sup> **3**, which forms an activated complex, **4**, with the triamide of phosphorous acid, **1**, **2** by electrophilic orientation on nitrogen (Scheme 4). As a result, one of the amido groups in the

SCHEME 4 For 1: R = CH<sub>3</sub>; 2: R = C<sub>2</sub>H<sub>5</sub>.

triamide, **1**, **2** is converted to a good leaving group and could then be easily substituted by nucleophilic attack on phosphorous. We have established by preliminary model studies that the quasi-phosphonium derivatives themselves do not possess a phosphorylation ability under the experimental conditions given.

Basic compounds such as pyridine, triethylamine, diethylamine etc. considerably decrease the rate of phosphorylation.

## EXPERIMENTAL

The tris-(N,N-dimethyl)-amide of phosphorous acid, **1**, and the tris-(N,N-diethyl)-amide of phosphorous acid, **2**, were prepared according to ref.<sup>10,11</sup> 1-O-Stearoylthane-2-ol, **a**, was prepared according to Reference 12. Dibenzoylperoxide (Merck) was used as a 10% benzene solution and dried over molecular sieve 3 Å. All other reagents were GR, or of purity in excess of 98% (Merck). Solvents were dried prior to use. Reaction conditions were kept strictly anhydrous.

Preparative thin-layer chromatography (TLC) was performed on 20 × 20 cm plates and stationary phase of silica gel G (Merck) with a layer thickness of 2 mm; after 1 h activation at 120°C. The following mobile phases were used: chloroform (system A); n-hexane/diethyl ether 95/5 (system B); n-hexane/chloroform 80/20, v/v (system C).

The melting points were determined on a Kofler melting point apparatus and are uncorrected.

<sup>13</sup>C NMR spectra were recorded on a Bruker AM-400 spectrometer at 100.62 MHz. <sup>13</sup>C chemical shifts are reported in ppm relative to tetramethylsilane (TMS). <sup>31</sup>P NMR spectra were recorded on a Bruker WP-80 spectrometer at 32.44 MHz. <sup>31</sup>P chemical shifts are reported in ppm relative to 85% phosphoric acid (external), where a positive sign is downfield from the standard. IR spectra were recorded on a Perkin-Elmer 337 spectrometer. Peak positions are reported in cm<sup>-1</sup>.

Elemental analyses were performed by the Microanalytical Service Laboratory, University of Plovdiv.

**1-O-Stearoylthane-2-O-bis(N,N-diethylamido)-thiophosphate, 2aS**; *Typical procedure.* A mixture of iodine (0.025 g; 0.1 mmol) and the tris-(N,N-diethyl)-amide of phosphorous acid (**2**; 0.519 g; 2.1 mmol) in benzene (50 mL) was heated at 75°C in a stream of argon for about 15 min until the precipitate dissolved, **2**.\* The solution was cooled to room temperature (20–25°C), 1-O-stearoylthane-2-ol (**a**; 0.657 g; 2 mmol) was added, and the reaction system was kept under these conditions for 40 min (The same procedure was also carried out at 75°C for about 1 min.). Then sulphur (0.067 g; 2.1 mmol) was added and the mixture was heated at 40°C for 5 min to give the thiophosphate derivative, **2aS**. Part of the solvent was removed under vacuum and the compound was isolated by TLC using system A as mobile phase, and then diethyl ether for eluting the product from the silica gel adsorbent.

Yield of **2aS**: 1.03 g (96%); n<sub>D</sub><sup>20</sup>: 1.4615; R<sub>f</sub>: 0.62 (system A). C<sub>28</sub>H<sub>50</sub>N<sub>2</sub>O<sub>3</sub>PS calc. C 62.86 H 11.14 N 5.24 P 5.79 S 6.00 (535.0) found C 62.57 H 11.12 N 5.20 P 5.83 S 6.04 <sup>13</sup>C NMR – {<sup>1</sup>H}(C<sub>6</sub>D<sub>6</sub>): δ = 14.2 (s, C-18); 23.3 (s, C-17); 25.7 (s, C-3); 30.0–30.6 (m, C-4–C-15); 32.7 (s, C-16); 34.6 (s, C-2); 63.6 (d, OCH<sub>2</sub>CH<sub>2</sub>OP, J = 10.0 Hz); 64.5 (s, OCH<sub>2</sub>CH<sub>2</sub>OP); 173.2 (s, C-1); 1-O-stearoylthane-2-O-fragment; 14.9 (s, CH<sub>3</sub>CH<sub>2</sub>N); 40.7 ppm (s, CH<sub>2</sub>N). <sup>31</sup>P NMR – {<sup>1</sup>H}(C<sub>6</sub>D<sub>6</sub>): δ = 77.7 ppm (s). IR (KBr): ν = 1750 (C=O); 1020, 790 (PO-C, P-OC); 720 (P-N); 695 cm<sup>-1</sup> (P=S).

**Cholesteryl-3-O-bis(N,N-diethylamido)-thiophosphate, 2bS.** Using cholesterol (**b**; 0.773 g; 2 mmol), the derivative was synthesized at room temperature (4 h) or at 75°C (6 min), and then purified (system B) the same way as described for **2aS**.

Yield of **2bS**: 1.09 g (92%); m.p. 73.0–74.5°C (from diethyl ether); R<sub>f</sub>: 0.52 (system B).

$C_{35}H_{65}N_2OPS$  calc. C 70.87 H 11.07 N 4.72 P 5.23 S 5.41 (593.1) found C 70.55 H 11.13 N 4.77 P 5.25 S 5.46  $^{13}C$  NMR –  $\{^1H\}(C_6D_6)$ :  $\delta$  = 12.5 (s, C-18); 19.6 (s, C-21); 19.9 (s, C-19); 21.9 (s, C-11); 23.2 (s, C-26); 23.5 (s, C-27); 24.8 (s, C-23); 25.1 (s, C-15); 28.8 (s, C-25); 29.1 (s, C-16); 30.8 (d, C-2,  $J$  = 4.4 Hz); 32.7 (s, C-7, C-8); 36.7 (s, C-20); 37.2 (s, C-22); 37.3 (s, C-10); 37.8 (s, C-1); 40.4 (s, C-24); 40.8 (s, C-12); 41.2 (d, C-4,  $J$  = 5.2 Hz); 43.1 (s, C-13); 50.8 (s, C-9); 57.1 (s, C-17); 57.5 (s, C-14); 76.6 (d, C-3,  $J$  = 3.7 Hz); 123.3 (s, C-6); 140.9 (s, C-5): cholesteryl-3-O-fragment; 14.7 (d,  $CH_3CH_2N$ ,  $J$  = 3.7 Hz); 41.0 ppm (d,  $CH_2N$ ,  $J$  = 3.7 Hz).  $^{31}P$  NMR –  $\{^1H\}(C_6D_6)$ :  $\delta$  = 77.2 ppm (s). IR (KBr):  $\nu$  = 1010, 790 (PO-C, P-OC); 730 (P-N); 695  $cm^{-1}$  (P=S).

*R,S*- $\alpha$ -Tocopheryl-6-O-bis(*N,N*-diethylamido)-thiophosphate, **2dS**. Using *R,S*- $\alpha$ -tocopherol (**d**; 0.861 g; 2 mmol), the derivative was synthesized within 10 h at room temperature (or for 40 min at 75°C) and then purified (system A) as described for **2aS**.

Yield of **2dS**: 1.24 g (98%);  $n_D^{20}$ : 1.5137; Rf: 0.90 (system A).  $C_{37}H_{69}N_2O_2PS$  calc. C 69.74 H 10.94 N 4.40 P 4.87 S 5.03 (637.2) found C 69.53 H 10.90 N 4.35 P 4.88 S 5.00  $^{13}C$  NMR –  $\{^1H\}(C_6D_6)$ :  $\delta$  = 12.8 (s,  $CH_3$ -5); 15.3 (s,  $CH_3$ -7); 15.8 (s,  $CH_3$ -8); 23.3 (s,  $CH_3$ -2); 32.2 (s, C-3); 41.0 (s, C-4); 75.4 (s, C-2); 118.3 (s, C-5, C-10); 123.6 (s, C-7, C-8); 144.1 (d, C-9,  $J$  = 8.9 Hz); 149.3 (s, C-6): nucleus; 20.4 (s,  $CH_3$ -4,  $CH_3$ -8); 21.7 (s, C-13); 21.9 (s,  $CH_3$ -12); 24.4 (d, C-2,  $J$  = 11.1 Hz); 25.4 (s, C-10); 25.7 (s, C-6); 28.8 (s, C-12); 33.6 (s, C-4); 33.7 (s, C-8); 38.4 (s, C-3, C-5, C-7, C-9); 40.3 (s, C-1, C-11): chain, *R,S*- $\alpha$ -tocopheryl-6-O-fragment; 14.5 (d,  $CH_3CH_2N$ ,  $J$  = 4.2 Hz); 41.6 ppm (d,  $CH_2N$ ,  $J$  = 4.2 Hz).  $^{31}P$  NMR –  $\{^1H\}(C_6D_6)$ :  $\delta$  = 72.6 ppm (s). IR (KBr):  $\nu$  = 1250, 840 (PO-C, P-OC<sub>aryl</sub>); 725 (P-N); 690  $cm^{-1}$  (P=S).

*Bis*(cholesteryl-3-O-)-(*N,N*-diethylamido)-phosphate, **2b<sup>2</sup>O**; *Typical procedure*. To a solution of the reagent, **2\***, prepared as described for **2aS**, cholesterol (**b**; 0.773 g; 2 mmol) was added and the reaction system was kept in a stream of argon at room temperature for 4 h. The same quantity of cholesterol (**b**; 0.773 g; 2 mmol) was then added, and the mixture was left under the same conditions for 6 days. Dibenzoylperoxide (0.509 g; 2.1 mmol) was added dropwise at 10°C to give the phosphate derivative, **2b<sup>2</sup>O**. Part of the solvent was removed under vacuum and the compound was crystallized from benzene/acetone.

Yield of **2b<sup>2</sup>O**: 1.48 g (83%); m.p. 147–149°C (from benzene/acetone), Rf: 0.43 (system A).  $C_{58}H_{100}NO_3P$  calc. C 78.19 H 11.35 N 1.57 P 3.49 (889.6) found C 78.22 H 11.30 N 1.57 P 3.43  $^{13}C$  NMR –  $\{^1H\}(C_6D_6)$ : The spectrum is very similar to that of **2bS**.  $^{31}P$  NMR –  $\{^1H\}(C_6D_6)$ :  $\delta$  = –2.0 ppm (s). IR (KBr):  $\nu$  = 1250 (P=O); 1015, 795 (PO-C, P-OC); 730  $cm^{-1}$  (P-N).

*Cholesteryl-3-O-(R,S- $\alpha$ -tocopheryl-6-O-)-(*N,N*-diethylamido)-thiophosphate, **2bdS**. Using **2\***, cholesterol (**b**; 0.773 g; 2 mmol) (first step: 75°C; 6 min) and *R,S*- $\alpha$ -tocopherol (**d**; 0.861 g; 2 mmol) (second step: 75°C; 4 h).*

The transformation to thiophosphate, **2bdS**, was done with sulphur (0.067 g; 2.1 mmol) at the temperature indicated. The compound was isolated by TLC using system C as mobile phase, and then diethyl ether for eluting the product from the silica gel adsorbent.

Yield of **2bdS**: 1.79 g (94%);  $n_D^{20}$ : 1.5149; Rf: 0.44 (system C).  $C_{60}H_{104}NO_3PS$  calc. C 75.79 H 11.05 N 1.47 P 3.26 S 3.38 (950.8) found C 75.85 H 11.00 N 1.44 P 3.30 S 3.41  $^{13}C$  NMR –  $\{^1H\}(C_6D_6)$ :  $\delta$  = 12.7 (s, C-18); 19.9 (s, C-21); 30.6 (s, C-2); 37.9 (d, C-1,  $J$  = 17.2 Hz): cholesteryl-3-O-fragment; *R,S*- $\alpha$ -tocopheryl-6-O-fragment: 23.4 (d,  $CH_3$ -2,  $J$  = 16.0 Hz); 32.2 (s, C-3): nucleus; 20.5 (d,  $CH_3$ -4,  $J$  = 10.0 Hz); 20.6 (d,  $CH_3$ -8,  $J$  = 10.0 Hz): chain; 15.0 (s,  $CH_3CH_2N$ ); 40.8 ppm (d,  $CH_2N$ ,  $J$  = 7.5 Hz).  $^{31}P$  NMR –  $\{^1H\}(C_6D_6)$ :  $\delta$  = 70.3 ppm (s). IR (KBr):  $\nu$  = 1245, 840 (PO-C, P-OC<sub>aryl</sub>); 1010, 790 (PO-C, P-OC); 720 (P-N); 685  $cm^{-1}$  (P = S).

*1-O-Stearoylthane-2-O-(R,S- $\alpha$ -tocopheryl-6-O-)-(*N,N*-diethylamido)-phosphate, **2adO**. Using **2\***, 1-O-stearoylthane-2-ol (**a**; 0.657 g; 2 mmol) (first step: 20–25°C; 40 min) and *R,S*- $\alpha$ -tocopherol (**d**; 0.861 g; 2 mmol) (second step: 65°C; 5 h).*

The transformation to phosphate, **2adO**, was done with dibenzoylperoxide (0.509 g; 2.1 mmol) at 10°C. The compound was isolated by TLC (system A) as described for **2bdS**.

Yield of **2adO**: 1.61 g (92%);  $n_D^{20}$ : 1.4672; Rf: 0.24 (system A).  $C_{53}H_{98}NO_6P$  calc. C 72.62 N 11.29 N 1.60 P 3.54 (876.5) found C 72.72 H 11.20 N 1.61 P 3.50  $^{13}C$  NMR –  $\{^1H\}(C_6D_6)$ :  $\delta$  = 14.2 (s, C-18); 63.6 (d,  $OCH_2CH_2OP$ ,  $J$  = 10.0 Hz); 64.5 (s,  $OCH_2CH_2OP$ ); 173.2 (s, C-1): 1-O-stearoylthane-2-O-fragment; *R,S*- $\alpha$ -tocopheryl-6-O-fragment: 12.7 (s,  $CH_3$ -5); 23.4 (d,  $CH_3$ -2,  $J$  = 16.0 Hz): nucleus; 20.3 (s,  $CH_3$ -4,  $CH_3$ -8); 24.3 (d, C-2,  $J$  = 16.0 Hz): chain; 14.9 (s,  $CH_3CH_2N$ ); 40.7 ppm (s,  $CH_2N$ ).  $^{31}P$  NMR –  $\{^1H\}(C_6D_6)$ :  $\delta$  = –0.2 ppm (s). IR (KBr):  $\nu$  = 1740 (C=O); 1250, 840 (PO-C, P-OC<sub>aryl</sub>); 1030, 790 (PO-C, P-OC); 725  $cm^{-1}$  (P-N).

*Bis*( $\beta$ -sitosteryl-3-O-)-(*N,N*-diethylamido)-phosphate, **2c<sup>2</sup>O**. A mixture of the reagent, **2\***, and

$\beta$ -sitosterol (c; 1.659 g; 4 mmol) in benzene (50 mL) was heated in a stream of argon at 75°C for 1 h. The procedures for the transformation to phosphate and the isolation of compound, **2c<sup>2</sup>O**, are identical with those described for **2b<sup>2</sup>O**.

Yield of **2c<sup>2</sup>O**: 1.76 g (93%); m.p. 140–145°C (from benzene/acetone); Rf: 0.45 (system A).  $C_{62}H_{108}NO_3P$  calc. C 78.66 H 11.52 N 1.48 P 3.27 (946.7) found C 78.72 H 11.55 N 1.50 P 3.24  $^{13}C$  NMR –  $\{^1H\}(C_6D_6)$ :  $\delta$  = 12.5 (s, C-18); 12.7 (s, C-29); 19.4 (s, C-26); 19.6 (s, C-27); 19.8 (s, C-21); 19.9 (s, C-19); 21.9 (s, C-11); 24.1 (s, C-28); 25.1 (s, C-15); 27.4 (s, C-23); 29.1 (s, C-16); 30.3 (s, C-25); 31.0 (s, C-2); 32.7 (s, C-7, C-8); 35.0 (s, C-22); 37.1 (s, C-10, C-20); 37.9 (s, C-1); 40.7 (s, C-12); 41.5 (s, C-4); 43.1 (s, C-13); 46.9 (s, C-24); 50.9 (s, C-9); 57.0 (s, C-17); 57.5 (s, C-14); 77.1 (s, C-3); 123.2 (s, C-6); 140.9 (s, C-5); bis( $\beta$ -sitosteryl-3-O-)-fragment; 14.8 (s,  $CH_3CH_2N$ ); 40.7 ppm (s,  $CH_2N$ ).  $^{31}P$  NMR –  $\{^1H\}(C_6D_6)$ :  $\delta$  = –2.1 ppm(s). IR (KBr):  $\nu$  = 1250 (P=O); 1010, 795 (PO-C, P-OC); 730  $cm^{-1}$  (P-N).

*Bis(cholesteryl-3-O-)-phosphite, 1b<sup>2</sup>OH*. A mixture of iodine (0.025 g; 0.1 mmol) and the tris-(N,N-dimethyl)-amide of phosphorous acid (**1**; 0.343 g; 2.1 mmol) in benzene (50 mL) was heated at 75°C in a stream of argon for about 20 min until the precipitate dissolved, **1\***. Cholesterol (b; 1.547 g; 4 mmol) was added and the system was kept under the same conditions for 30 min. The transformation to phosphite, **1b<sup>2</sup>OH**, was accomplished with acetic acid (0.150 g; 2.5 mmol) for 1 h at the temperature indicated, whereupon part of the solvent was removed under vacuum and the compound was crystallized from benzene/acetone.

Yield of **1b<sup>2</sup>OH**: 1.51 g (92%); m.p. 176–177°C (from benzene/acetone); Rf: 0.51 (system A).  $C_{54}H_{91}O_3P$  calc. C 79.14 H 11.22 P 3.78 (819.5) found C 79.06 H 11.25 P 3.70  $^{13}C$  NMR –  $\{^1H\}(C_6D_6)$ :  $\delta$  = 12.5 (s, C-18); 19.5 (s, C-21); 31.0 (d, C-2,  $J$  = 21.7 Hz); 41.4 (d, C-4,  $J$  = 17.2 Hz); 140.1 ppm (s, C-5).  $^{31}P$  NMR ( $C_6D_6$ ):  $\delta$  = 2.6 ppm (d,  $J_{P-H}$  = 671.6 Hz). IR (KBr):  $\nu$  = 2390 (PH); 1260 (PO); 1010, 798  $cm^{-1}$  (PO-C, P-OC).

*Bis( $\beta$ -sitosteryl-3-O-)-phosphite, 1c<sup>2</sup>OH*. Using  $\beta$ -sitosterol (c; 1.659 g; 4 mmol), the derivative, **1c<sup>2</sup>OH**, was synthesized and isolated as described for **1b<sup>2</sup>OH**.

Yield of **1c<sup>2</sup>OH**: 1.59 g (91%); m.p. 165–168°C (from benzene/acetone); Rf: 0.53 (system A).  $C_{58}H_{99}O_3P$  calc. C 79.56 H 11.42 P 3.54 (875.6) found C 79.48 H 11.45 P 3.50  $^{13}C$  NMR –  $\{^1H\}(C_6D_6)$ :  $\delta$  = 12.5 (s, C-18); 12.7 (s, C-29); 19.7 (s, C-19, C-21); 31.0 (d, C-2,  $J$  = 17.2 Hz); 41.4 (d, C-4,  $J$  = 17.2 Hz); 140.3 ppm (s, C-5).  $^{31}P$  NMR ( $C_6D_6$ ):  $\delta$  = 2.7 ppm (d,  $J_{P-H}$  = 671.6 Hz). IR (KBr):  $\nu$  = 2390 (PH); 1260 (PO); 1015, 795  $cm^{-1}$  (PO-C, P-OC).

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## REFERENCES AND NOTES

1. H. G. Khorana, *Some Recent Developments in the Chemistry of Phosphate Esters of Biological Interest* (Wiley, New York, 1961).
2. R. Burgada, *Ann. Chim.* **1**, 15 (1966); *C.A.* **65**, 3904 c (1966).
3. L. A. Slotin, *Synthesis*, 737 (1977).
4. K. A. Petrov, E. E. Nifant'ev, T. N. Lysenko and V. P. Evdakov, *Zh. Obshch. Khim.* **31**, 2377 (1961); *J. Gen. Chem. USSR*, **31**, 2214 (1961).
5. E. E. Nifant'ev and D. A. Predvoditelev, *Bioorg. Khim.* **7**, 1285 (1981); *C.A.* **95**, 187533 (1981).
6. E. E. Nifant'ev, D. A. Predvoditelev, A. P. Tuseev, M. K. Grachev and M. A. Zolotov, *Zh. Obshch. Khim.* **50**, 1702 (1980); *J. Gen. Chem. USSR*, **50**, 1379 (1980).
7. M. A. Zolotov, D. A. Predvoditelev and E. E. Nifant'ev, *Zh. Obshch. Khim.* **50**, 2380 (1980); *C.A.* **94**, 24247 (1981).
8. R. Burgada, *Ann. Chim.* **8**, 347 (1963); *C.A.* **59**, 15164 f (1963).
9. H. Nöth and H. J. Vetter, *Chem. Ber.* **94**, 1505 (1961).
10. C. Stuebe and H. P. Lankelma, *J. Am. Chem. Soc.* **78**, 976 (1956).
11. A. B. Burg and P. J. Slota, *J. Am. Chem. Soc.* **80**, 1107 (1958).
12. H. Eibl and O. Westphal, *Liebigs Ann. Chem.* **709**, 244 (1967).